Year 4: Looking at States

Subject Specific Vocabulary		Working Scientifically	By the end of this unit, I will know:
water vapour	Water that is in the form of gas.	Asking relevant questions and using different types of scientific	Solids - you can hold it in your hands. You can form it into a pile. It is not easy to change the
condensation	When water vapour that is around us changes from a gas back to liquid.	enquiries to answer them Setting up simple practical	shape of a material in the solid state.
precipitation	Any watery substance such as rain, water, snow, hail or sleet that falls to Earth.	 enquiries, comparative and fair tests Recording findings using simple scientific language, drawings, labelled diagrams, keys, bar charts and tables 	Liquids - you cannot hold it in your hands. It forms a pool, not a pile. Liquids take the shape of the bottom of the container they are in.
evaporation	When liquid changes into gas, usually when it heats up.		Gas - a material escapes from an unsealed
substance	Any solid, liquid, powder or gas is a substance.	Identifying similarities, differences or changes related to simple scientific ideas and	container. It spreads out to fill all the space available, and takes the shape of the entire container.
matter	Any solid, liquid or gas that exists in the universe.	processes Using straightforward scientific	Melting – solid turns into a liquid, because
melting	When heat is applied to a solid the particles vibrate more. This causes the	evidence to answer questions	heat has been applied.
	particles to move further apart, breaking the solid into a liquid.	or to support their findings.	Freezing - the reverse of melting - from liquid to solid.
heating	When a solid is heated it gains energy. And the particles move more. This causes a change of state.		Evaporation – liquid turns into a gas because heat has been applied. It can happen at any temperature.
boiling	To heat or become heated to the temperature at which bubbles form in a liquid and rise to the top	Vetiting Freeding Subinnation Liquid Condentation Gas	Condensation – gas turns to liquid through cooling.
solidify	The cooling of a liquid slows the particles and they become solid at or near room temperature	Roleg Toleg Toleg Toleg Coud formation Transpiration	The Water Cycle describes the path that all water follows as it moves around Earth in different states. It involves evaporation and condensation.
freezing	Similar to solidifying, but at very cold temperatures.		Our famous scientist for this unit is: Bettye Washington Greene
solid	A substance that stays the same shape. Its particles do not move.	Contentiation	
liquid	Liquids will flow as they are made up of loosely packed particles.	Russet	
gas	Gaseous matter is made up of matter that is so loose it is always moving.		